

- relevance QM
- LA & QM

relevance - de Broglie relation

Einstein: $E = \hbar \omega = \frac{\hbar}{2\pi} 2\pi \nu = \hbar \nu$ - photon energy depends on its frequency (colour)

also $p = \hbar k = \frac{h}{\lambda}$ from Planck's law of radiation

$E = pc$ (relativity with $m=0$)

$\lambda \nu = c$ (# oscillations \times length of 1 oscillation)

$\hbar \omega = pc$

$\hbar \omega = p \lambda \nu$

$\frac{h}{2\pi} 2\pi \nu = p \lambda \nu \Rightarrow p = \frac{h}{\lambda}$

de Broglie: $p = \frac{h}{\lambda}$ holds in general

$\lambda = \frac{h}{p}$ $\leftarrow 6.626 \cdot 10^{-34} \text{ m}^2 \text{ kg}^{-1} \text{ s}^{-1}$; $\lambda = \frac{h}{\sqrt{2mE}}$ $E = \frac{p^2}{2m} \rightarrow p = \sqrt{2mE}$

1) me crashing from bicycle at 10 m/s

$p = mv = 800 \frac{\text{kg} \cdot \text{m}}{\text{s}}$

$\lambda = \frac{6.626 \cdot 10^{-34}}{800} = 8.28 \cdot 10^{-37} \text{ m}$

2) C_{60} molecule \rightarrow mass = $60 \cdot 12.01 \cdot 1.66 \cdot 10^{-27} \text{ kg} \approx 1.2 \cdot 10^{-24} \text{ kg}$
(60) carbon atomic mass unit

$v = 200 \text{ m/s}$

$\Rightarrow p = 2.4 \cdot 10^{-22} \text{ kg m/s}$

$\lambda = \frac{6.626 \cdot 10^{-34}}{2.4 \cdot 10^{-22}} = 2.76 \cdot 10^{-12} \text{ m}$ ($\lambda = \text{pm}$)

Atadt et al Nature 401 680 (1999)

3) electron ~~from~~ from ~~100~~ 1 eV

\Rightarrow atomic units : $E = [\text{Ha}]$, $[m] = m_e$, $[\hbar] = 1 = \frac{e^2}{4\pi\epsilon_0}$

$1 \text{ eV} \approx \frac{1}{27.2} \text{ Ha}$

$\lambda = \frac{2\pi \hbar}{\sqrt{2mE}} \stackrel{\text{a.u.}}{=} \frac{2\pi}{\sqrt{2E}} = \frac{2\pi}{\sqrt{13.6}} = 2\pi \cdot 3.7 = 23.2 \text{ \AA}$

what? Bohr

1 Bohr = 0.529 \AA

what? Angstrom

OK :) $23.25 \approx 12.5 \text{ \AA} = 1.25 \cdot 10^{-9} \text{ m}$

4) free particle with temperature T

$$E = \frac{3}{2} k_B T ; k_B = 1,3806 \cdot 10^{-23} \text{ J K}^{-1}$$

→ to a.u.

$$1 \text{ eV} = 1,602 \cdot 10^{-19} \text{ J}$$

$$1 \text{ eV} = \frac{1}{27,2} \text{ Ha}$$

$$\left. \begin{array}{l} 1 \text{ eV} = 1,602 \cdot 10^{-19} \text{ J} \\ 1 \text{ eV} = \frac{1}{27,2} \text{ Ha} \end{array} \right\} \begin{array}{l} 1,602 \cdot 10^{-19} \text{ J} = \frac{1}{27,2} \text{ Ha} \\ 43,57 \cdot 10^{-19} \text{ J} = \text{Ha} \end{array}$$

$$k_B = 1,3806 \cdot 10^{-23} \text{ J K}^{-1}$$

$$J = 2,295 \cdot 10^{17} \text{ Ha}$$

$$= 1,3806 \cdot 10^{-23} \cdot 2,295 \cdot 10^{17} \text{ Ha K}^{-1}$$

$$= 3,1685 \cdot 10^{-6} \text{ Ha K}^{-1}$$

$$\rightarrow E = \frac{3}{2} \cdot 3,1685 \cdot 10^{-6} \text{ Ha K}^{-1} \cdot T \quad [\text{H ionisation} \approx \frac{1}{2} \text{ Ha} \Rightarrow 10^5 \text{ K}]$$

(is really $\sim 10^4 \text{ K}$ only...)

$$= 4,75275 \cdot 10^{-6} \cdot T$$

$$p = \sqrt{2mE} = \sqrt{2 \cdot 4,75275 \cdot 10^{-6} \text{ Ha} \cdot T} = 3,083 \cdot 10^{-3} \sqrt{\text{Ha} \cdot T}$$

EE) neutron $T = 110 \text{ K}$; $m_n = 1839 m_e$

$$p = 3,083 \cdot 10^{-3} \sqrt{\text{Ha} \cdot T} = 3,083 \cdot 10^{-3} \sqrt{1839 \cdot 0,11} = 0,0438$$

$$\lambda = \frac{2\pi \hbar}{p} \approx \frac{2\pi}{0,0438} = 143,3 \text{ a.u.}$$

1) ~~what is temperature~~

1) proton 20K:

$$\lambda = \frac{2\pi \hbar}{\sqrt{2m_p E}} = \frac{2\pi \hbar}{\sqrt{2m_p \frac{3}{2} k_B T}} = \frac{2\pi}{3,083 \cdot 10^{-3} \sqrt{\text{Ha} \cdot T}} = \frac{2\pi}{3,083 \cdot 10^{-3} \sqrt{1836 \cdot 20}}$$

$$= 10,65 \approx 5,62 \text{ \AA}$$

2) proton 300K:

$$\lambda = \frac{2\pi \hbar}{\sqrt{2m_p E}} = \frac{2\pi}{3,083 \cdot 10^{-3} \sqrt{1836 \cdot 300}} = 2,745 = 1,45 \text{ \AA}$$

~ size of H-bond
 P don't drink heavy water?
 too much

3) deuterium 300K

$$\lambda = \lambda_{\text{proton}} \cdot \frac{1}{\sqrt{2}} \approx 1 \text{ \AA}$$

Operatory

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$$\frac{d}{dx}(f+g) = f' + g'$$

$$\frac{d}{dx} \frac{1}{f} : O(f+g) = \frac{1}{f+g}$$

$$\frac{d}{dx}(cf) = c \frac{df}{dx}$$

$$O(f) = \frac{1}{f} \quad O(g) = \frac{1}{g}$$

$$O(f) + O(g) \neq O(f+g)$$

$$\frac{1}{\sqrt{\pi}} \sin(n\phi), \frac{1}{\sqrt{\pi}} \cos(n\phi), \frac{1}{\sqrt{2\pi}} ; (0, 2\pi), n \in \mathbb{N}$$

normalizacja:

$$f = \frac{1}{\sqrt{\pi}} \sin(n\phi)$$

$$\int_0^{2\pi} f^* f d\phi = \frac{1}{\pi} \int_0^{2\pi} \sin^2(n\phi) d\phi$$

$$= \frac{1}{\pi} \int_0^{2\pi} \frac{1}{(2i)^2} [e^{in\phi} - e^{-in\phi}]^2 d\phi$$

$$= \frac{1}{\pi} \int_0^{2\pi} \left(-\frac{1}{4}\right) [e^{2in\phi} + e^{-2in\phi} - 2] d\phi$$

$$= \frac{1}{\pi} \int_0^{2\pi} \frac{1}{2} d\phi = \frac{1}{2\pi} \cdot 2\pi = 1 \quad \text{OK}$$

$$f = \frac{1}{\sqrt{\pi}} \sin(n\phi)$$

$$g = \frac{1}{\sqrt{\pi}} \cos(n\phi)$$

$$\int_0^{2\pi} f^* g d\phi = \frac{1}{\pi} \int_0^{2\pi} \sin(n\phi) \cos(n\phi) d\phi$$

$$= \frac{1}{\pi} \frac{1}{4i} \int_0^{2\pi} [e^{in\phi} - e^{-in\phi}] [e^{in\phi} + e^{-in\phi}] d\phi$$

$$= \frac{1}{4\pi i} \int_0^{2\pi} e^{2in\phi} - e^{-2in\phi} d\phi = 0 \quad \text{OK}$$

$$\hat{A} = i \frac{d}{d\phi}$$

$$\hat{A} \frac{1}{\sqrt{\pi}} \sin \phi = \frac{1}{\sqrt{\pi}} \cos \phi \quad \hat{A} \frac{1}{\sqrt{\pi}} \sin \phi = -\frac{1}{\sqrt{\pi}} \cos \phi \quad \hat{A} \frac{1}{\sqrt{2\pi}} = 0$$

$$\hat{B} = \sin \phi$$

$$\hat{B} \frac{1}{\sqrt{\pi}} \sin \phi = \frac{1}{\sqrt{\pi}} \sin^2 \phi = \frac{1}{\sqrt{\pi}} \text{Re} \left[\frac{1}{2i} (e^{i\phi} - e^{-i\phi}) \right]^2 = \frac{1}{\sqrt{\pi}} \frac{1}{-4} [e^{2i\phi} + e^{-2i\phi} - 2]$$

$$= \frac{1}{\sqrt{\pi}} \left[\frac{1}{2} - \frac{1}{2} \cos(2\phi) \right] \leftarrow \text{kombinacja 2-foai, to je OK}$$

(niech do ulasni'stwa)

$$\hat{B} \frac{1}{\sqrt{\pi}} \cos \phi = \frac{1}{\sqrt{\pi}} \cos \phi \sin \phi = \frac{1}{\sqrt{\pi}} \frac{1}{4i} (e^{i\phi} + e^{-i\phi})(e^{i\phi} - e^{-i\phi}) = \frac{1}{4i\sqrt{\pi}} (e^{2i\phi} - e^{-2i\phi} + 0)$$

$$= \frac{1}{4i\sqrt{\pi}} \frac{1}{2\sqrt{\pi}} \sin(2\phi)$$

$$\hat{B} \frac{1}{\sqrt{2\pi}} = \frac{1}{\sqrt{2\pi}} \sin \phi$$